

ELIMINATION OF 14 $\beta$ -HYDROXY-12-MESYLOXY-5 $\beta$ -STEROIDS

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**Abstract:** Elimination of a 12 $\alpha$ -methanesulfoxy group in 3 $\beta$ -acetoxy-14 $\beta$ -hydroxy-12 $\alpha$ -mesyloxy-5 $\beta$ -androstane-17 $\beta$ -carboxylic-acid-14 $\beta$ -lactone under preventing Wagner-Meerwein rearrangements by steric fixation, is reported. Furthermore, the synthesis of this compound and the different elimination behaviors in the 12 $\alpha$ - and the 12 $\beta$ -isomers are described.

Elimination of 12-sulfoxy-groups in steroids for introduction of 11(12)-double-bonds is an important procedure in organic syntheses<sup>1</sup>. The major products, however, are Wagner-Meerwein rearranged products<sup>2</sup>. Reactions of 12 $\beta$ -mesyloxy- or 12 $\beta$ -tosyloxy steroids, especially in 5 $\beta$ ,14 $\beta$ -hydroxy-cardenolides and pregnanes gave high yields of rearranged products<sup>3</sup>. Instead of developing appropriate reagents and conditions for the introduction of the double-bond, we tried to modify the steroid in order to prevent undesired side reactions. We would now like to report that this rearrangement can be suppressed, if the reaction is carried out with the 14 $\beta$ ,20-lactone, where the D-ring is fixed in a rigid conformation. The correspondent 12-mesyloxy,14 $\beta$ ,20-lactones were synthesized in the following way (scheme I):

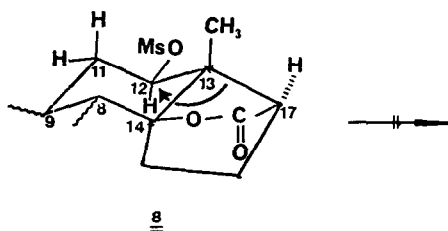
Oxidation of digoxin 1<sup>4</sup> using a modified procedure in DMF with CrO<sub>3</sub><sup>5</sup>, resulted, after 30h, in a yield of 60% in 2, mp 270-272°C. By acetylation to 3 and degradation of the side-chain with KMnO<sub>4</sub><sup>6</sup>, up to 52% of the 17 $\beta$ -carboxylic-acid 4 was obtained, mp 235-238°C. After stirring 4 in THF with a slight excess of carbonyldiimidazol (CDI) for 1h, an excess of NaH was added and allowed to react for 1 more hour to yield 85% of the 14 $\beta$ ,20-lactone 5, mp 176-178°C.

This cyclization was carried out first with 11-oxo-14 $\beta$ -hydroxy-20-carboxylic-acids using acetic anhydride/pyridine, by heating the reaction mixture under reflux<sup>7</sup>. Our conditions are smoother and can also be used for the lactonization of the 3 $\beta$ ,12 $\beta$ ,14 $\beta$ -trihydroxy-5 $\beta$ -androstane-17 $\beta$ -carboxylic-acid

to the 3 $\beta$ ,12 $\beta$ ,14 $\beta$ -trihydroxy-5 $\beta$ -androstane-17 $\beta$ -carboxylic-acid-14 $\beta$ -lactone without acetylation of the alcohols.

Reduction with NaBH<sub>4</sub> led to a mixture of the 12 $\alpha$ - and 12 $\beta$ -alcohols 6, mp 154-156°C and 7, mp 202-203°C. The yields of the 12 $\alpha$ -alcohol increased, depending upon the solvent, in the following sequence: butanol(27%), ethanol (33%), THF(70%). Sulfonation of the equatorial 12 $\beta$ -alcohol 7 in pyridine with methane-sulfonyl-chloride is achieved quantitatively within 30 min. to 8, mp 199-201°C (scheme II). After heating 8 in collidine under reflux for 40h, only the unchanged starting material was recovered. Sulfonation of the axial 12 $\alpha$ -alcohol 6 to 9 is achieved quantitatively with methane-sulfonyl-chloride/pyridine in 12h, mp 155-158°C. The increased reaction time is required, due to the steric hindrance of the axial hydroxy group. Elimination of 9 in collidine under reflux for 30h yielded 72% of the desired product 10, mp 180-181°C.

Obviously, it is possible to suppress the rearrangement<sup>3</sup> by steric fixation of the D-ring. Usually, the elimination of the sulfoxy-ester with following rearrangement is carried out with high yields within 1h. In the sterically fixed conformation of the 14 $\beta$ ,20-lactone, the C-17 carbon is not able to attack the C-12 position anymore:



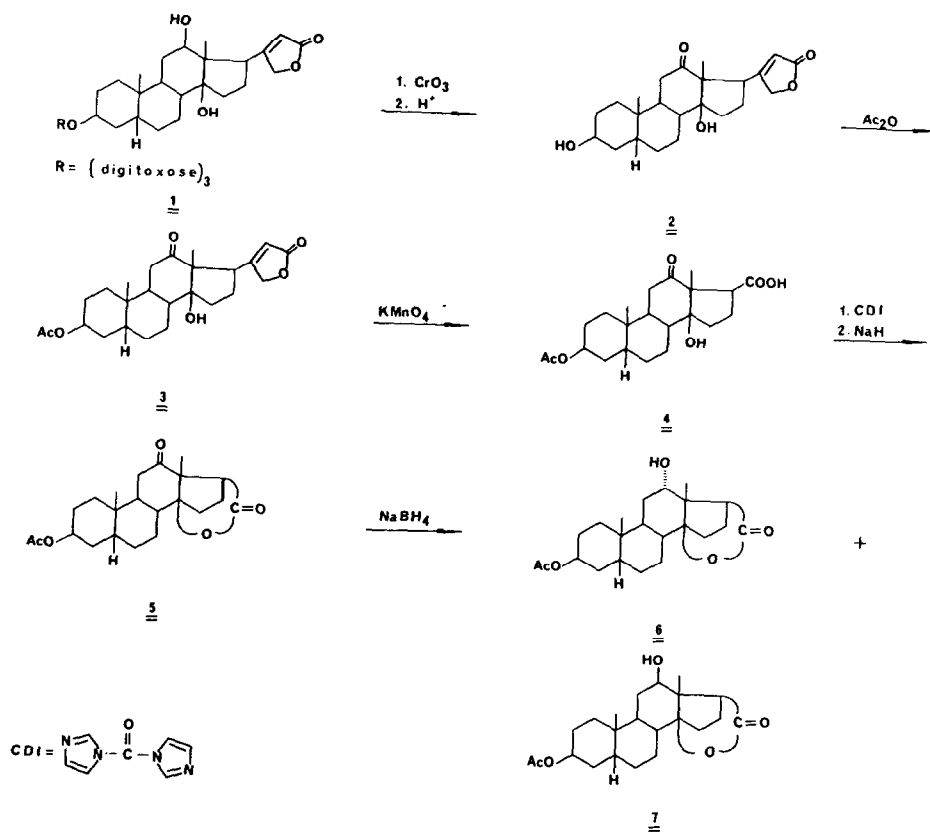
The prevented development of the 11(12)-double-bond in the 12 $\beta$ -mesyloxy compound 8 is due to the low elimination tendency in the trans-diequatorial position of the  $\beta$ -hydrogen and the sulfoxy-ester. In contrast to this observation, elimination of the 12 $\alpha$ -mesyloxy-group 9 is possible, because of the trans-diaxial positions of the leaving groups.

All the substances described, were characterized via C,H analysis and their NMR spectra<sup>8</sup>.

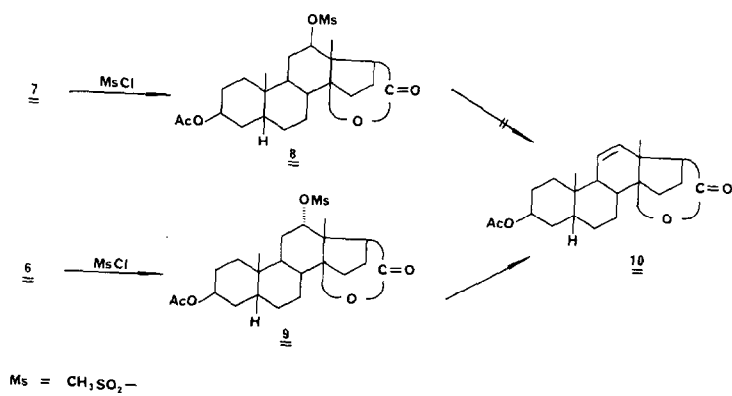
Further studies in using this valuable reaction for synthesis of 14 $\beta$ -hydroxy-steroids, which are known to possess biological activity, are in progress.

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Scheme I



Scheme II



References and Notes:

1. For example see: Ch.R. Engel and F. Papadopoulos, *J. Org. Chem.* 26, 2868 (1961); G. Just and Ch.R. Engel, *J. Org. Chem.* 23, 12 (1958); C.H. Chen, *Synthesis* 1976, 125; G.H. Possner, G.M. Gurria and K.A. Babiak, *J. Org. Chem.* 42, 3173 (1977).
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8. <sup>1</sup>H-NMR data (300 MHz, CDCl<sub>3</sub>): 4 δ=1.0(s,3H,H-19), 1.2(s,3H,H-18), 3.9(d,d,1H,J=4.5Hz,J=9Hz,H-17), 5.1(pseudo s,1H,H-3); 5 δ=1.1(s,3H,H-19) 1.25(s,3H,H-18), 2.95(pseudo d,1H,H-17), 5.1(pseudo s,1H,H-3), 6 δ=1.0(s,6H,H-18 and H-19), 2.05(s,3H,CH<sub>3</sub>COO), 2.8(pseudo d,1H,H-17), 4.1(pseudo s,1H,H-12), 5.1(pseudo s,1H,H-3); 7 δ=1.05(s,6H,H-18 and H-19), 2.05(s,3H,CH<sub>3</sub>COO), 2.8(pseudo d,1H,H-17), 3.75(d,d,1H,J=5Hz,J=11Hz, H-12), 5.1(pseudo s,1H,H-3); 8 δ=1.1(s,3H,H-19), 1.15(s,3H,H-18), 2.05(s,3H,CH<sub>3</sub>COO), 2.9(pseudo d,1H,H-17), 3.1(s,3H,CH<sub>3</sub>SO<sub>2</sub>), 4.85(pseudo d, 1H,H-12), 5.1(pseudo s,1H,H-3); 9 δ=1.0(s,3H,H-19), 1.15(s,3H,H-18), 2.05(s,3H,CH<sub>3</sub>COO), 2.88(pseudo d,1H,H-17), 3.05(s,3H,CH<sub>3</sub>SO<sub>2</sub>), 4.9 (pseudo s,1H,H-12), 5.1(pseudo s,1H,H-3); 10 δ=1.0(s,3H,H-19), 1.1(s, 3H,H-18), 2.05(s,3H,CH<sub>3</sub>COO), 2.7(pseudo d,1H,H-17), 5.1(pseudo s,1H, H-3), 5.5(d,1H,J=4Hz,H-12), 5.75(d,d,1H,J=4Hz,J=9Hz,H-11).

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